

## Two New Compounds from the Ascomycete *Daldinia concentrica*

by Xiang-Dong Qin<sup>a)b)</sup>, Ze-Jun Dong<sup>a)</sup>, and Ji-Kai Liu<sup>\*a)</sup>

<sup>a)</sup> State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, the Chinese Academy of Sciences, Kunming 650204, P. R. China

<sup>b)</sup> The Graduate School of the Chinese Academy of Science, Beijing 100039, P. R. China

---

Two new compounds, 1-isopropyl-2,7-dimethylnaphthalene (**1**) and 21-(acetyloxy)-6,13,14-trihydroxy-16,18-dimethyl-10-phenyl[11]cytochalasa-7,19-dien-1-one (**2**), were isolated from the fruiting bodies of *Daldinia concentrica*. The structures were established by spectroscopic methods.

---

**Introduction.** – *Daldinia concentrica* has aroused a great deal of scientific interest because of its unique secondary metabolites. Allport and Bu'Lock studied European and American *Daldinia* sp. in 1958 and 1960 [1][2], which resulted in the identification of characteristic metabolites in their stromata and cultures. Some of these compounds have antimicrobial and nematocidal activities [3]. During recent studies on *Daldinia* sp., more than 20 new metabolites have been discovered, including benzoquinones [4], a binaphthyl and benzophenones [5], cytochalasins [6–8], a daldiniapyrone and daldinones [9], heptenetriols [10], triterpenoids [9][11][12], and derivatives of azaphilone [13], of which some show a range of biological activities. A benzofuran lactone has recently been isolated that has been found to be active against HIV-1 *in vitro* [14]. More recently, we reported two aromatic steroids from *Daldinia concentrica* and proposed that the origin of these compounds is derived from the transformation undergone by their precursor due to microbial action. These two compounds are recognized to be the long-sought, biological precursor steroids for organic matter in Earth's subsurface [15].

In the present study, the structures of the new alkylnaphthalene **1** and the new cytochalasin **2** isolated from *Daldinia concentrica* are reported. Alkylnaphthalenes have been isolated from some plants [16–18] and some geological samples [19][20]. Their occurrence plays an important role in the study of the sedimentary process, and can also be used as a maturity parameter for some sediments and crude oils. Cytochalasins are a group of fungal secondary metabolites identified from diverse fungal sources which have a wide range of biological activities [21–23][6–8] but are best known for their various effects on mammalian [24]. The most unusual of their properties is their ability to cause cells to extrude their nuclei, leading to the formation of nuclei-free cells. At lower concentrations, they interfere with cell division by preventing cytoplasmic division leading to binuclear or polynuclear cells and also inhibit cell movement [6].

**Results and Discussion.** – Compound **1** was obtained as colorless oil. The molecular formula of **1** was deduced to be  $C_{15}H_{18}$  on the basis of a molecular-ion peak at  $m/z$  198 in a FAB-MS (positive mode) and its  $^{13}C$ -NMR spectrum (*Table 1*). Further spectral data allowed to elucidate the structure of compound **1** as 1-isopropyl-2,7-dimethylnaphthalene (*Fig. 1*).

Table 1.  $^1H$ - and  $^{13}C$ -NMR Spectral Data ( $CDCl_3$ ) of **1**.  $\delta$  in ppm,  $J$  in Hz.

	$\delta(C)$ (DEPT)	$\delta(H)$	$^1H, ^1H$ -COSY	$^{13}C, ^1H$ -HMBC (selected)
C(1)	142.1 (s)			H–C(3), H–C(8), Me–C(2), Me <sub>2</sub> CH
C(2)	131.8 (s)			Me–C(2), H–C(4)
H–C(3)	125.6 (d)	7.18 (d, $J=7.3$ )	H–C(4)	Me–C(2)
H–C(4)	121.4 (d)	7.25 (d, $J=7.3$ )	H–C(3)	
H–C(5)	124.8 (d)	7.87 (d, $J=8.8$ )	H–C(6)	
H–C(6)	127.2 (d)	7.30 (d, $J=8.8$ )	H–C(5), H–C(8)	Me–C(7)
C(7)	134.7 (s)			H–C(5), Me–C(7)
H–C(8)	122.9 (d)	7.90 (s)	H–C(6)	H–C(6), Me–C(7)
C(8a)	131.1 (s)			
C(4a)	131.5 (s)			H–C(3), H–C(6)
Me–C(2)	19.4 (q)	2.61 (s)		H–C(3)
Me–C(7)	22.0 (q)	2.52 (s)		H–C(8)
Me <sub>2</sub> CH	28.2 (d)	3.70 (m)	Me <sub>2</sub> CH	Me <sub>2</sub> CH
Me <sub>2</sub> CH	23.6 (q)	1.36 (d, $J=6.8$ )	Me <sub>2</sub> CH	Me <sub>2</sub> CH

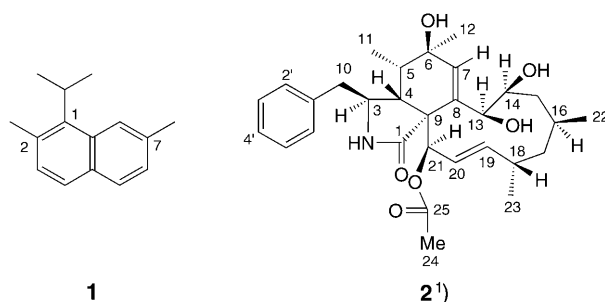


Fig. 1. Structures of compounds **1** and **2**

The  $^{13}C$ -NMR spectrum of **1** showed fifteen signals (4 Me, 6 CH, 5 C) including ten aromatic C-atoms. Considering the seven degrees of unsaturation, a naphthalene skeleton was deduced. In the  $^1H$ -NMR spectrum, a *d* was observed at  $\delta$  1.36 (6 H), suggesting the existence of an  $^iPr$  group. Thus, compound **1** was a naphthalene substituted by 2 Me and 1  $^iPr$  groups. Their location was determined by 2D-NMR. In the  $^1H, ^1H$ -COSY, the cross-peaks H–C(5)/H–C(6), H–C(6)/H–C(8), H–C(3)/H–C(4) allowed to place H–C(5), H–C(6), H–C(8) at one ring and H–C(3), and H–C(4) at the other ring of the naphthalene structure (*Table 1*). In the  $^1H, ^{13}C$ -HMBC spectrum, the correlations H–C(3), H–C(8), Me–C(2), and Me<sub>2</sub>CH/C(1), Me–C(2)/C(3), and Me–C(7)/C(6) and C(8) were observed, suggesting the position of an  $^iPr$  group at C(1) and of the Me groups at C(2) and C(7).

<sup>1)</sup> Trivial atom numbering; for the systematic name, see *Exper. Part*.

Table 2.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR Spectral Data ( $\text{CD}_3\text{OD}$ ) of **2**<sup>1</sup>.  $\delta$  in ppm,  $J$  in Hz.

	$\delta(\text{C})$ (DEPT)	$\delta(\text{H})$	$^1\text{H}, ^1\text{H}$ -COSY	$^{13}\text{C}, ^1\text{H}$ -HMBC (selected)
C(1)	178.5 (s)			H–C(3), H–C(21)
H–C(3)	55.6 ( <i>d</i> )	3.86 ( <i>m</i> )	H–C(4), CH <sub>2</sub> (10)	H–C(5)
H–C(4)	45.1 ( <i>d</i> )	2.95 ( <i>t</i> , $J=6.0$ )	H–C(3), H–C(5)	CH <sub>2</sub> (10), Me(11), H–C(21)
H–C(5)	39.7 ( <i>d</i> )	2.09 ( <i>m</i> )	H–C(4), Me(11)	H–C(3), H–C(7), Me(12)
C(6)	70.7 (s)			H–C(4), Me(11)
H–C(7)	142.2 ( <i>d</i> )	6.04 (s)		Me(12), H–C(13)
C(8)	135.9 (s)			H–C(4), H–C(14)
C(9)	55.4 (s)			H–C(7), H–C(13), H–C(20)
CH <sub>2</sub> (10)	45.4 ( <i>t</i> )	3.10 ( <i>dd</i> , $J=4.5, 13.6$ ), 2.81 ( <i>dd</i> , $J=8.1, 13.6$ )	H–C(3)	H–C(4), H–C(2',6')
Me(11)	12.7 ( <i>q</i> )	1.09 ( <i>d</i> , $J=7.2$ )	H–C(5)	H–C(4)
Me(12)	23.7 ( <i>q</i> )	1.19 (s)		H–C(5), H–C(7)
H–C(13)	74.1 ( <i>d</i> )	4.02 ( <i>d</i> , $J=8.8$ )	H–C(14)	H–C(7), CH <sub>2</sub> (15)
H–C(14)	71.4 ( <i>d</i> )	3.71 ( <i>br. t</i> , $J=8.8$ )	H–C(13), CH <sub>2</sub> (15)	
CH <sub>2</sub> (15)	46.8 ( <i>t</i> )	1.58 ( <i>dd</i> , $J=9.6, 13.8$ ), 1.16 ( <i>m</i> )	H–C(14), H–C(16)	H–C(13), CH <sub>2</sub> (17)
H–C(16)	29.7 ( <i>d</i> )	1.28 ( <i>m</i> )	CH <sub>2</sub> (15), CH <sub>2</sub> (17), Me(22)	H–C(14)
CH <sub>2</sub> (17)	46.4 ( <i>t</i> )	1.40 ( <i>dd</i> , $J=6.1, 13.5$ ), 0.84 ( <i>ddd</i> , $J=6.4, 8.8, 13.5$ )	H–(16), H–C(18)	CH <sub>2</sub> (15), H–C(19)
H–C(18)	40.9 ( <i>d</i> )	2.19 ( <i>m</i> )	CH <sub>2</sub> (17), H–C(19), Me(23)	H–C(20)
H–C(19)	147.6 ( <i>d</i> )	5.84 ( <i>dd</i> , $J=10.6, 14.8$ )	H–C(18), H–C(20)	CH <sub>2</sub> (17), H–C(21)
H–C(20)	123.6 ( <i>d</i> )	5.32 ( <i>dd</i> , $J=10.2, 14.8$ )	H–C(19), H–C(21)	H–C(18)
H–C(21)	76.7 ( <i>d</i> )	5.89 ( <i>d</i> , $J=10.2$ )	H–C(20)	H–C(4), H–C(19)
Me(22)	22.0 ( <i>q</i> )	0.93 ( <i>d</i> , $J=6.6$ )		CH <sub>2</sub> (15), CH <sub>2</sub> (17)
Me(23)	21.0 ( <i>q</i> )	0.96 ( <i>d</i> , $J=6.5$ )	H–C(18)	CH <sub>2</sub> (17), H–C(19)
Me(24)	21.5 ( <i>q</i> )	2.05 (s)		
C(25)	171.3 (s)			H–C(21)
C(1')	139.1 (s)			H–C(3), CH <sub>2</sub> (10), H–C(3',5')
H–C(2',6')	130.7 ( <i>d</i> )	7.26 ( <i>d</i> , $J=7.1$ )		CH <sub>2</sub> (10)
4-C(3',5')	129.9 ( <i>d</i> )	7.34 ( <i>t</i> , $J=7.5$ )		
H–C(4')	128.0 ( <i>d</i> )	7.25 ( <i>t</i> , $J=7.3$ )		

The molecular formula of compound **2** was determined to be  $\text{C}_{30}\text{H}_{41}\text{NO}_6$  on the basis of the HR-TOF-MS ( $m/z$  534.2818 ( $[M + \text{Na}]^+$ )). The structure of a [11]cytochalasadienone for **2** (see *Fig. 1*) was established by its spectral data (*Table 2*) and comparison with literature data.

The  $^{13}\text{C}$ -NMR spectrum of **2** ( $\text{CD}_3\text{OD}$ ) showed 28 resolved peaks corresponding to 28 C-atoms (*Table 2*), which were classified into 5 Me, 3 CH<sub>2</sub>, 4 CH, 4 NCH or OCH, 6  $\text{sp}^2$  CH, 2  $\text{sp}^3$  C, 2  $\text{sp}^2$  C, and 2 C=O groups by analysis of the DEPT spectra. The  $^1\text{H}$ -NMR spectrum of **2** displayed 37 proton signals (*Table 2*). Deduced from the molecular formula, the presence of 2 OH groups, one amino group, and

one Ac group was suggested. The connectivity of the proton and C-atoms was established by the  $^1\text{H}$ ,  $^{13}\text{C}$ -HMOC spectrum. Analysis of the  $^1\text{H}$ ,  $^1\text{H}$ -COSY plot revealed the five partial structures **I–V** (Fig. 2).  $^{13}\text{C}$ ,  $^1\text{H}$  Long-range couplings ( $^3J$ ) observed in the  $^1\text{H}$ ,  $^{13}\text{C}$ -HMBC experiment gave the following evidence: 1) The cross-peaks from H–C(3) ( $\delta$  3.86) to C–C(1') ( $\delta$  139.1), from  $\text{CH}_2$ –C(10) ( $\delta$  3.10, 2.81) to C(2',6') ( $\delta$  130.7), from H–C(3',5') ( $\delta$  7.34) to C–C(1') showed the presence of a benzyl moiety connecting with partial structure **I**). 2) The cross-peaks from H–C(4) ( $\delta$  2.95) to C(6) ( $\delta$  70.7), C(8) ( $\delta$  135.9), C(10) ( $\delta$  45.4), C(11) ( $\delta$  12.7), and C(21) ( $\delta$  76.7), from H–C(5) ( $\delta$  2.09) to C(3) ( $\delta$  55.6) and C(12) ( $\delta$  23.7), from H–C(7) ( $\delta$  6.04) to C(5) ( $\delta$  39.7), C(9) ( $\delta$  55.4), and C(12), from Me(11) ( $\delta$  1.09) to C(4) ( $\delta$  45.1), and from Me(12) ( $\delta$  1.19) to C(5) ( $\delta$  39.7) and C(7) ( $\delta$  142.2) indicated a 4,5,6,8,9-pentasubstituted<sup>1)</sup> cyclohexene containing the partial structure **I**, **III**, and **IV**. 3) The long-range couplings from H–C(3) and H–C(4) to C(1) ( $\delta$  178.5) suggested that a pyrrole ring was attached to the cyclohexene ring. 4) The long-range couplings from H–C(7) to C(13) ( $\delta$  74.1), from H–C(13) ( $\delta$  4.02) to C(7) and C(9), and from H–C(14) ( $\delta$  3.71) to C(8) suggested that partial structure **II** was attached to the cyclohexene ring. 5) The long-range couplings from H–C(13) to C(15) ( $\delta$  46.8), from H–C(14) to C(16) ( $\delta$  29.7), from  $\text{CH}_2$ (15) ( $\delta$  1.58, 1.16) to C(13), C(17) ( $\delta$  46.4), and C(22) ( $\delta$  22.0), from  $\text{CH}_2$ (17) ( $\delta$  1.40, 0.84) to C(15), C(19) ( $\delta$  147.6), C(22), and C(23) (21.0), from H–C(18) ( $\delta$  2.19) to C(20) ( $\delta$  123.6), from H–C(19) ( $\delta$  5.84) to C(17), C(21), and C(23), from H–C(20) ( $\delta$  5.32) to C(9) and C(18) ( $\delta$  40.9), from H–C(21) to C(1) ( $\delta$  178.5), C(4), C(19), and C(25) ( $\delta$  171.3) suggested the connection of partial structures **I–V** and the pyrrole ring. The  $J(\text{H,H})$  was 14.8 Hz between H–C(19) ( $\delta$  5.84) and H–C(20) ( $\delta$  5.32), indicating that the olefin moiety has the (*E*)-configuration. The  $\delta(\text{C})$  74.1, 71.4, and 76.7 supported the presence of 2 OH and 1 Ac group. Finally, the cytochalasan skeleton was suggested because of the degree of unsaturation and the molecular formula of **2**.

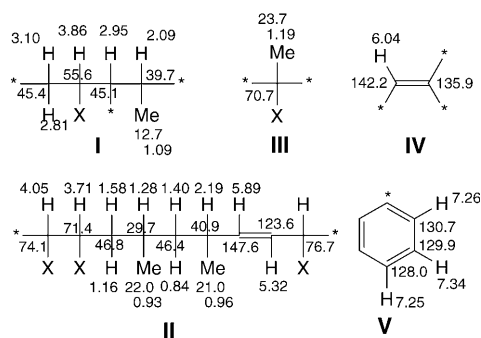


Fig. 2. Partial structures **I** to **V** of compound **2**

The relative configuration of **2** was established by NOESY experiments (Table 3). The NOEs H–C(4)/ $\text{CH}_2$ (10) and H–C(3)/Me(11), indicated the  $\beta$ -configurations for H–C(4) and H–C(5), and an  $\alpha$ -configuration for H–C(3) [25][26]. No NOEs between Me(12) and H–C(4) and H–C(5) revealed an  $\alpha$  configuration for Me(12) [27]. H–C(7) displayed NOEs to Me(12), H–C(13), and H–C(14), and

Table 3. NOEs of compound **2**

H	NOE at H	H	NOE at H
H–C(3)	$\text{CH}_2$ (10), Me(11)	H–C(19)	$\text{CH}_2$ (15), Me(23)
H–C(4)	H–C(5), $\text{CH}_2$ (10), Me(11), H–C(20)	H–C(20)	H–C(4), H–C(18)
H–C(5)	H–C(4)	H–C(21)	H–C(13), H–C(14)
H–C(7)	Me(12) H–C(13), H–C(14)	Me(22)	$\text{CH}_2$ (15), $\text{CH}_2$ (17), H–C(18)
H–C(13)	H–C(7), $\text{CH}_2$ (15), $\text{CH}_2$ (17), H–C(21)	Me(23)	H–C(16), $\text{CH}_2$ (17), H–C(19)
H–C(14)	H–C(7), $\text{CH}_2$ (15), $\text{CH}_2$ (17), H–C(21)		

H–C(21) displayed NOEs to H–C(13) and H–C(14), indicating that H–C(13), H–C(14), and H–C(21) were on the  $\alpha$  side of the 11-membered ring. On the other hand, NOEs from H–C(4)/H–C(20), H–C(20)/H–C(18), H–C(18)/Me(22), and Me(23)/H–C(16) required H–C(18), H–C(20), and Me(22) to be on the  $\beta$  side of the ring [25][26].

We wish to acknowledge financial support from the *National Natural Science Foundation of China* (30470027 and 30225048).

### Experimental Part

**General.** CC = Column chromatography. M.p.: XRC-1 apparatus (Sichuan University, Sichuan, People's Republic of China). Optical rotation; *Horiba-SEPA-300* automatic polarimeter (*Horiba*, Tokyo, Japan). UV Spectra: *Shimadzu-UV2401PC* spectrophotometer (*Shimadzu Corporation*, Kyoto, Japan);  $\lambda_{\max}$  in nm. IR Spectra: *Bruker Tensor-27* spectrophotometer (*Bruker*, Karlsruhe, Germany); in  $\text{cm}^{-1}$ . 1D and 2D NMR Spectra: *Bruker DRX-500* NMR instruments (*Bruker*, Karlsruhe, Germany); at 500 MHz for  $^1\text{H}$  and 125 MHz for  $^{13}\text{C}$ ;  $\delta$  in ppm rel. to  $\text{SiMe}_4$  as an internal standard, coupling constants  $J$  in Hz. Mass Spectra: *VG-Autospec-3000* mass spectrometer (*VG*, Manchester, England) and *API Qstar Pulsar* (*Applied Biosystems*, Foster City, USA); in  $m/z$  (rel.%).

**Mushroom Material.** Fruiting bodies of *Daldinia concentrica* were collected in Laojunshan, Yunnan, P. R. China, in 2003. The voucher specimen was deposited at the herbarium of Kunming Institute of Botany, the Chinese Academy of Sciences.

**Extraction and Isolation.** Dried and powdered fruiting bodies (11.5 kg, dry weight) were extracted with  $\text{CHCl}_3$  ( $4 \times 50$  l) at r.t. The extract (340 g) was subjected to CC (silica gel,  $\text{CHCl}_3/\text{MeOH}$  100:0, 95:5, 9:1 (v/v)): *Fractions 1–21*. *Fr. 1* (eluted with  $\text{CHCl}_3$ ; 1 g) was separated by CC (silica gel, petroleum ether/acetone 100:0, 98:2, 95:5, 9:1, 8:2): *Fr. 1.1–1.22*. *Fr. 1.4* (eluted with petroleum ether) was purified by prep. TLC (petroleum ether/hexane 45:1): **1** (41.0 mg). *Fr. 10* (eluted with  $\text{CHCl}_3$ ; 0.6 g) was separated by CC (silica gel,  $\text{CHCl}_3/\text{MeOH}$  100:0, 95:5, 9:1). *Fr. 10.1–10.7*. *Fr. 10.5* (eluted with  $\text{CHCl}_3/\text{MeOH}$  95:5; 0.1 g) was purified by CC (*Sephadex LH-20*): **2** (11.2 mg).

**1-Isopropyl-2,7-dimethylnaphthalene (1):** Colorless oil. UV ( $\text{CHCl}_3$ ): 240, 292. IR: 3064, 3008, 2962, 2925, 2868, 1626, 1601, 1460, 1442, 1382, 1362, 831, 807, 778. NMR: see *Table 1*. FAB-MS (pos.): 198 (60,  $M^+$ ).

**21-(Acetyloxy)-6,13,14-trihydroxy-16,18-dimethyl-10-phenyl[11]cytochalasa-7,19-dien-1-one (= (3S, 3aR, 4S, 5R, 7R, 8S, 10S, 12S, 15R, 15aR)-15-(Acetyloxy)-2,3,3a,4,5,7,8,9,10,11,12,15-dodecahydro-5,7,8-trihydroxy-4,5,10,12-tetramethyl-3-(phenylmethyl)-1H-cycloundec[d]isoindol-1-one; 2):** Colorless needles. M.p. 197–199°.  $[\alpha]_{\text{D}}^{24} = +51.17$  (MeOH,  $c = 1.3$ ). UV (MeOH): 204. IR: 3426, 3028, 2953, 2926, 1743, 1687, 1496, 1453, 1371, 1234, 1102, 1016, 956, 702. NMR: see *Table 2*. FAB-MS (neg.): 510 (100,  $[M - H]^-$ ). HR-TOF-MS: 534.2818 ( $[M + Na]^+$ ,  $\text{C}_{30}\text{H}_{41}\text{NNaO}_6^+$ ; calc. 534.2831).

### REFERENCES

- [1] D. C. Allport, J. D. Bu'Lock, *J. Chem. Soc.* **1958**, 4090.
- [2] D. C. Allport, J. D. Bu'Lock, *J. Chem. Soc.* **1960**, 654.
- [3] H. Anke, M. Stadler, A. Mayer, O. Sterner, *Can. J. Bot.* **1995**, *73*, 802.
- [4] X. D. Qin, J. K. Liu, *Helv. Chim. Acta* **2004**, *87*, 2022.
- [5] T. Hashimoto, S. Tahara, S. Takaoka, T. Tori, *Chem. Pharm. Bull.* **1994**, *42*, 1528.
- [6] M. S. Buchanan, T. Hashimoto, Y. Asakawa, *Phytochemistry* **1995**, *40*, 135.
- [7] M. S. Buchanan, T. Hashimoto, S. Takaoka, Y. Kan, Y. Asakawa, *Phytochemistry* **1996**, *42*, 173.
- [8] M. S. Buchanan, T. Hashimoto, Y. Asakawa, *Phytochemistry* **1996**, *41*, 821.
- [9] D. N. Quang, T. Hashimoto, M. Tanaka, M. Baumgartner, M. Stadler, Y. Asakawa, *J. Nat. Prod.* **2002**, *65*, 1869.
- [10] F. Wang, J. K. Liu, *Helv. Chim. Acta* **2004**, *87*, 2131.
- [11] D. N. Quang, T. Hashimoto, M. Tanaka, M. Baumgartner, M. Stadler, Y. Asakawa, *Phytochemistry* **2002**, *61*, 345.

- [12] M. Stadler, M. Baumgartner, T. Grothe, A. Muehlbauer, S. Seip, H. Wollweber, *Phytochemistry* **2001**, *56*, 787.
- [13] T. Hashimoto, S. Tahara, S. Takaoka, M. Tori, *Chem. Pharm. Bull.* **1994**, *42*, 2397.
- [14] X. D. Qin, Z. J. Dong, J. K. Liu, L. M. Yang, R. R. Wang, Y. T. Zheng, Y. Lu, Y. S. Wu, Q. T. Zheng, *Helv. Chim. Acta* **2006**, *89*, 127.
- [15] X. D. Qin, J. K. Liu, *J. Nat. Prod.* **2004**, *67*, 2133.
- [16] M. C. Sriraman, B. A. Nagasampagi, R. C. Pandey, S. Dev, *Tetrahedron* **1973**, *29*, 985.
- [17] B. Van Dooren, R. Bos, D. H. E. Tattje, *Planta Med.* **1981**, *42*, 385.
- [18] T. Kajiwara, S. Ochi, K. Kodama, K. Matsui, A. Hatanaka, T. Fujimura, T. Ikeda, *Phytochemistry* **1992**, *31*, 783.
- [19] R. Alexander, R. I. Kagi, R. K. Singh, I. B. Sosrowidjojo, *Org. Geochem.* **1994**, *21*, 115.
- [20] T. P. Bastow, R. Alexander, S. J. Fisher, R. K. Singh, B. G. K. van Aarssen, R. I. Kagi, *Org. Geochem.* **2000**, *31*, 523.
- [21] W. G. Thilly, H. L. Liber, G. N. Wogan, in 'Cytochalasins, Biochemical and Cell Biological Aspects' Ed. S. W. Tanenbaum, Elsevier/North-Holland Biochemical Press, Amsterdam, 1978.
- [22] S. Natori, I. Yahara, in 'Cytochalasins in Mycotoxins and Phytoalexins', Eds. R. P. Shama and D. K. Salunkhe, CRC Press, Boca Raton, 1991, p. 291–336.
- [23] Y. Izawa, T. Hirose, T. Shimizu, K. Koyama, S. Natori, *Tetrahedron* **1989**, *45*, 2323.
- [24] S. B. Cater, *Nature (London)* **1967**, *213*, 261.
- [25] T. Tomikawa, K. Shin-Ya, T. Kinoshita, A. Miyajima, H. Seto, Y. Hayakawa, *J. Antibiot.* **2001**, *54*, 379.
- [26] T. Tomikawa, K. Shin-Ya, T. Kinoshita, A. Miyajima, H. Seto, Y. Hayakawa, *J. Antibiot.* **2002**, *55*, 666.
- [27] Y. Izawa, T. Hirose, T. Shimizu, K. Koyama, S. Natori, *J. Antibiot.* **1989**, *45*, 2323.

Received October 21, 2005